

The Need for Standardized Benchmarking of SOC_s

10.10.2023 | CHRISTIAN LENSER AND NORBERT H. MENZLER

INSTITUTE OF ENERGY AND CLIMATE RESEARCH – MATERIALS SYNTHESIS AND PROCESSING (IEK-1)

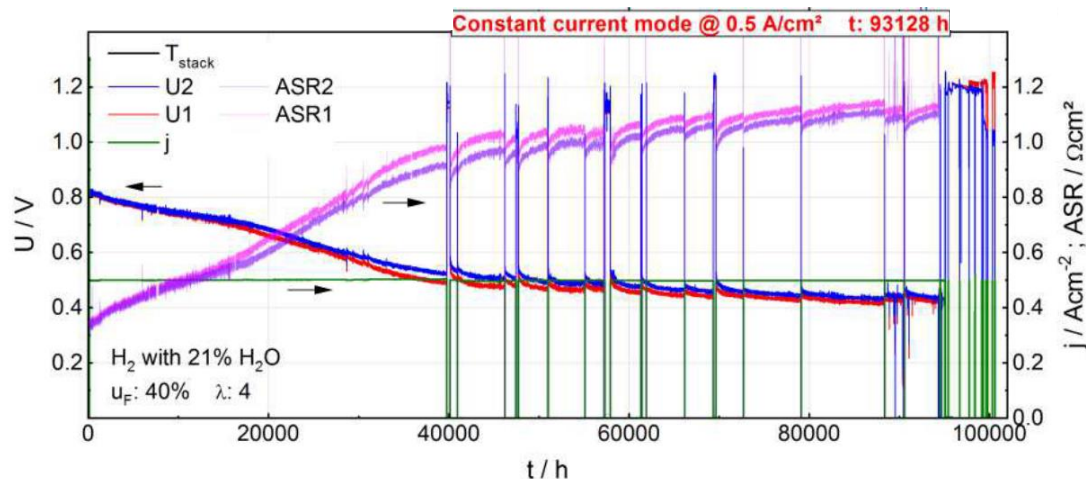
FORSCHUNGSZENTRUM JÜLICH, GERMANY

Overview

- Benchmarking Solid Oxide Cells (SOCs) – status quo in literature
- Problems with the status quo
- Can we have better metrics?

SOC research at Forschungszentrum Jülich

- SOC research for > 25 years
- From raw materials to stacks and systems
- Cell, interconnect, sealing, BoP, design & engineering



Q. Fang, L. Blum, D. Stolten, ECS Transactions, 91 (2019) 687-696.

Member of the Helmholtz Association



10/40 kW rSOC system

- Highlights include degradation testing of SOFC / SOEC, rSOC system design and testing, and much more...

SOC – R&D status and industrialization

- SOC technology is on the cusp of large-scale commercialization (MW to GW)
 - Fuel cell: Bosch (DE), Weichai (CN), Bloom Energy (USA), Elcogen (ES), SolydEra (IT), Convion (FI)...
 - Electrolysers: Bloom Energy (USA), Haldor Topsoe (DN), Sunfire (DE),...
- As the SOC's move toward a focus on production, there is a need for industry and key players to identify promising new technology and materials
- Cell testing is the gold standard to assess materials and cell design, but there some problems with how cell testing is reported

Motivation

Performance Benchmark of Planar Solid Oxide Cells Based on Material Development and Designs

David Udomsilp, Christian Lenser, Olivier Guillon,* and Norbert H. Menzler**

- Review article about the state-of-the-art in SOC
- Searched for current density (A cm^{-2}) at 0.7 V
- Quick reference for cell / materials performance
- Major obstacle: the way that cell performance is reported

600 °C		700 °C	
$j_{0.7\text{V}}$ [A cm ⁻²]	R_p [Ω cm ²]	$j_{0.7\text{V}}$ [A cm ⁻²]	R_p [Ω cm ²]

[1] D. Udomsilp, C. Lenser, O. Guillon, N.H. Menzler, Energy Technology, 9 (2021) 2001062.

Benchmarking SOFC – the status quo

- Searching Web of Science (results in last 5 years, „solid oxide fuel cell“, „electrode“)
- Papers focusing on materials development use the peak power density (PPD) as metric

Application of CuNi–CeO₂ fuel electrode in oxygen electrode

Code: 3D Core–Shell-Structured $\text{La}_2\text{O}_3\text{--}\delta\text{@Pr}_2\text{O}_3\text{--}\delta$ Nanofibers Prepared by Coaxial

Boosting and Robust Multifunction Fuel Cells

supported reversible solid oxide cell

International Journal of Hydrogen

ACS APPLIED
ENERGY MATERIALS

Ultrafine, Dual-Phase, Cation-Deficient $\text{PrBa}_{0.8}\text{Ca}_{0.2}\text{Co}_2\text{O}_{5+\delta}$ Air Electrode for Efficient Solid Oxide Cells

nanocomposite electrodes

Journal

ACS APPLIED MATERIALS & INTERFACES

ADVANCED
ENERGY
MATERIALS

as for Reversible Enabled by an

Sr-free orthorhombic perovskite
 $\text{Pr}_{0.8}\text{Ca}_{0.2}\text{Fe}_{0.8}\text{Co}_{0.2}\text{O}_{3-\delta}$ as a high-
performance air electrode for
reversible solid oxide cell

Journal of Power Sources

High-Performance $\text{SmBaMn}_2\text{O}_{5+\delta}$ Electrode for Solid Oxide Fuel Cell

The logo for ChemistMater, featuring the letters 'cm' in a bold, lowercase, sans-serif font, followed by the words 'CHEMIST' and 'MATER' stacked vertically in a smaller, uppercase, sans-serif font.

High Cu content $\text{LaNi}_{1-x}\text{Cu}_x\text{O}_{3-\delta}$

perovskites as candidate air electrode materials for Reversible Solid Oxide Electrode for Solid oxide

Cells

Review of SOFC Cathode Performance Enhancement by Surface Modifications: Recent Advances and Future Directions

International Journal of Hydrogen

Energy

Ni₂Co₂SnO₈ Compounds as Cathode Materials for Solid Oxide Fuel Cells**

energy&fuels

Angewandte
Zuschriften 

Availability of data

Information supplied in main paper

Reference	Area	Gas composition/ flow rates	Contacting materials	Current sweep rate
A				
B				
C				
D				
E				
F				

[A] Z. Yue, L. Jiang, Z. Chen, N. Ai, Y. Zou, S.P. Jiang, C. Guan, X. Wang, Y. Shao, H. Fang, Y. Luo, K. Chen, ACS Applied Materials & Interfaces, 15 (2023) 8138-8148.

[B] J. Bai, D. Zhou, X. Zhu, N. Wang, R. Chen, B. Wang, ACS Applied Energy Materials, 5 (2022) 11178-11190.

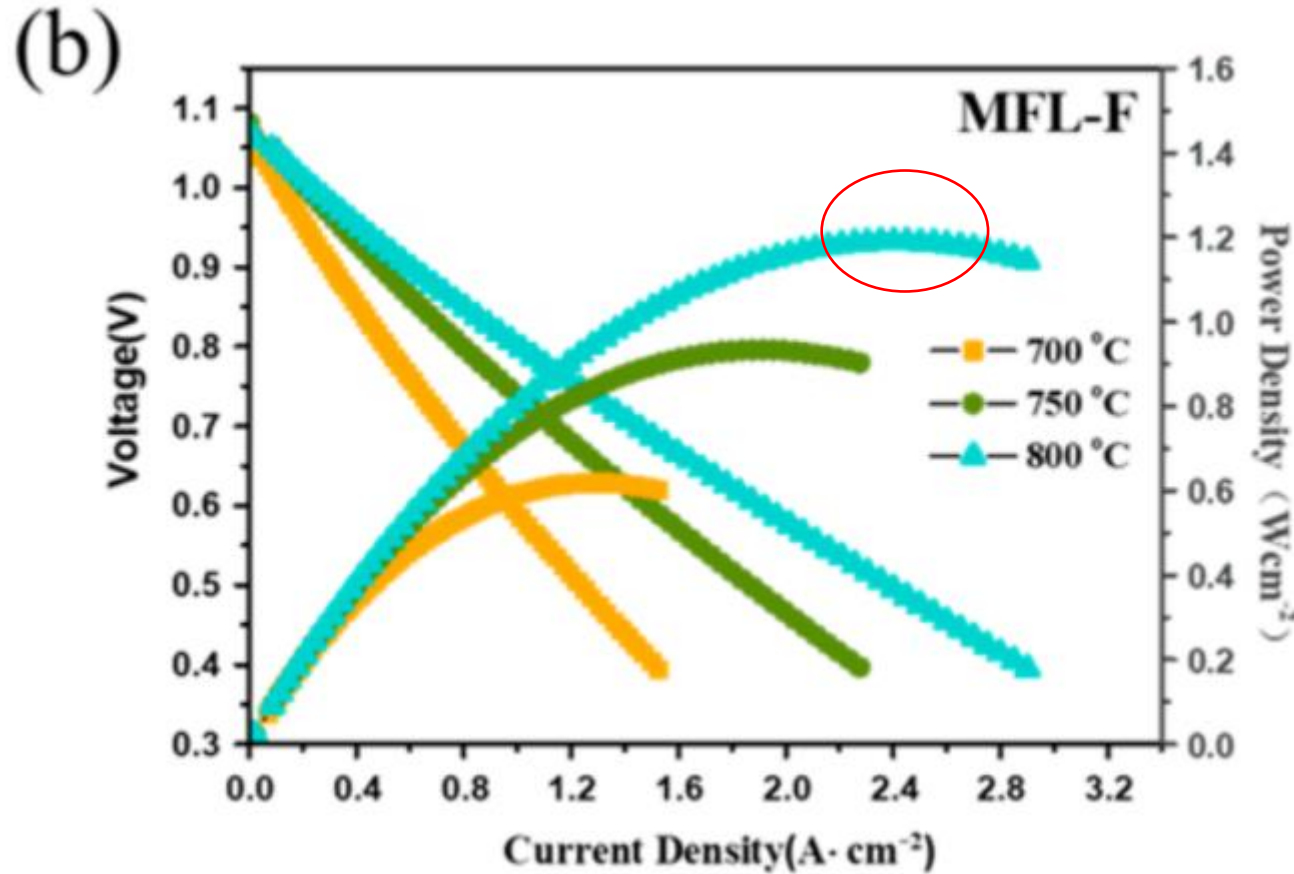
[C] Z. Zheng, J. Jing, H. Yu, Z. Yang, C. Jin, F. Chen, S. Peng, ACS Sustainable Chemistry & Engineering, 10 (2022) 6817-6825.

[D] S. Yoo, A. Jun, Y.-W. Ju, D. Odhhuu, J. Hyodo, Y. Jeong Hu, N. Park, J. Shin, T. Ishihara, G. Kim, Angewandte Chemie International Edition, 53 (2014) 13064-13067.

[E] J. Zamudio-García, L. dos Santos-Gómez, J.M. Porras-Vázquez, E.R. Losilla, D. Marrero-López, Journal of the European Ceramic Society, 43 (2023) 1548-1558.

[F] Y. Niu, Y. Zhou, W. Zhang, Y. Zhang, C. Evans, Z. Luo, N. Kane, Y. Ding, Y. Chen, X. Guo, W. Lv, M. Liu, Adv. Energy Mater., 12 (2022) 2103783.

What is the peak power density (PPD)?



- PPD is the maximum of the power density vs current density curve
- PPD provides one value that describes cell performance at a given temperature
-or does it?

Z. Zheng, J. Jing, H. Yu, Z. Yang, C. Jin, F. Chen, S. Peng, ACS Sustainable Chemistry & Engineering, 10 (2022) 6817-6825.

PPD – the problems

Reference	Temperature (°C)	Voltage (V)	Current density (A cm ⁻²)	Power density (W cm ⁻²)
A	700	0.47	1.3	0.61
B	700	0.5	2.3	1.15
C	700	0.55	0.4	0.22

- PPD is never at the same working point for different cells
- PPD is always at rather low voltages
 - Consequences for efficiency
 - Unstable operation

PPD – voltage efficiency

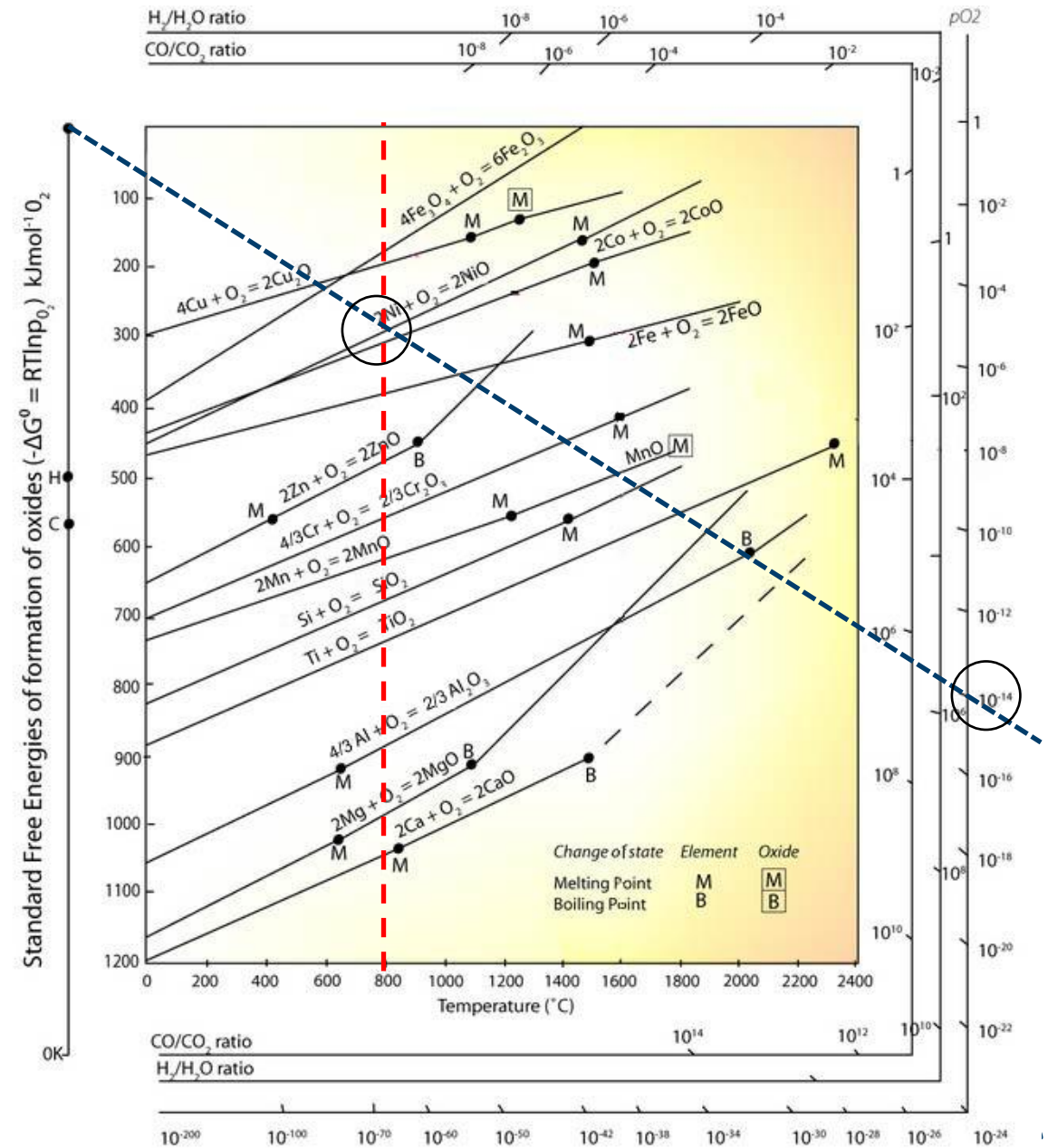
- Fuel cells operating below the thermoneutral voltage $V_{TN} = \frac{-\Delta H}{nF}$ generate heat
- As $V_{TN} \sim 1.3 \text{ V}$ at $800 \text{ }^{\circ}\text{C}$, SOFCs always produce heat
- The further V_{cell} is below V_{TN} , the more heat (and less electricity) is produced

- Definition:

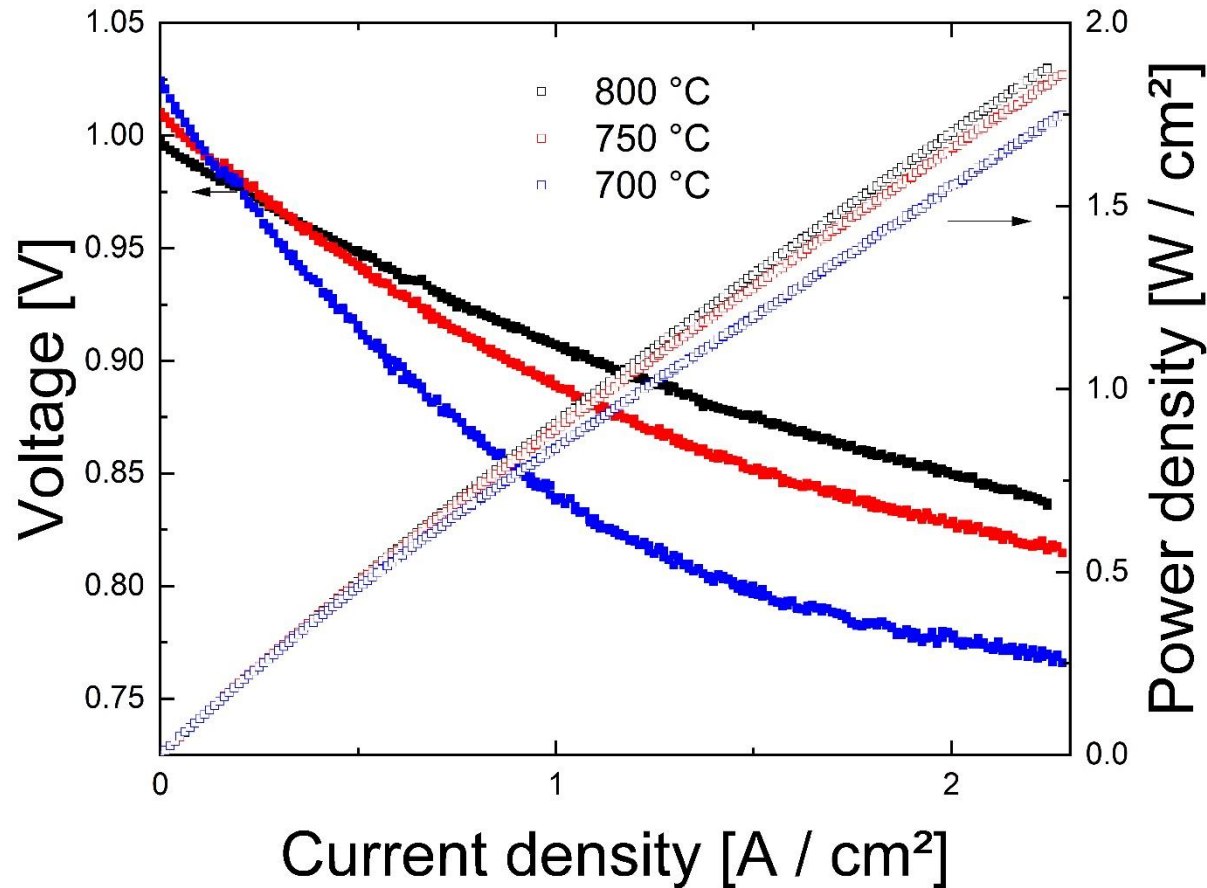
$$\text{Voltage efficiency} = \frac{\text{Operating voltage (V)}}{\text{Thermodynamic voltage (E)}}$$

PPD – low voltage

- Ellingham diagram: equilibrium pO_2 for Ni oxidation is $\sim 10^{-14}$ atm at 800°C
- With $E = -\frac{RT}{4F} \ln\left(\frac{pO_2}{0.21}\right)$; this pO_2 corresponds to a cell voltage of 700 mV at 800°C
- From this simple analysis, it should be clear the PPD at ~ 500 mV does not reflect a stable operation point of the SOFC



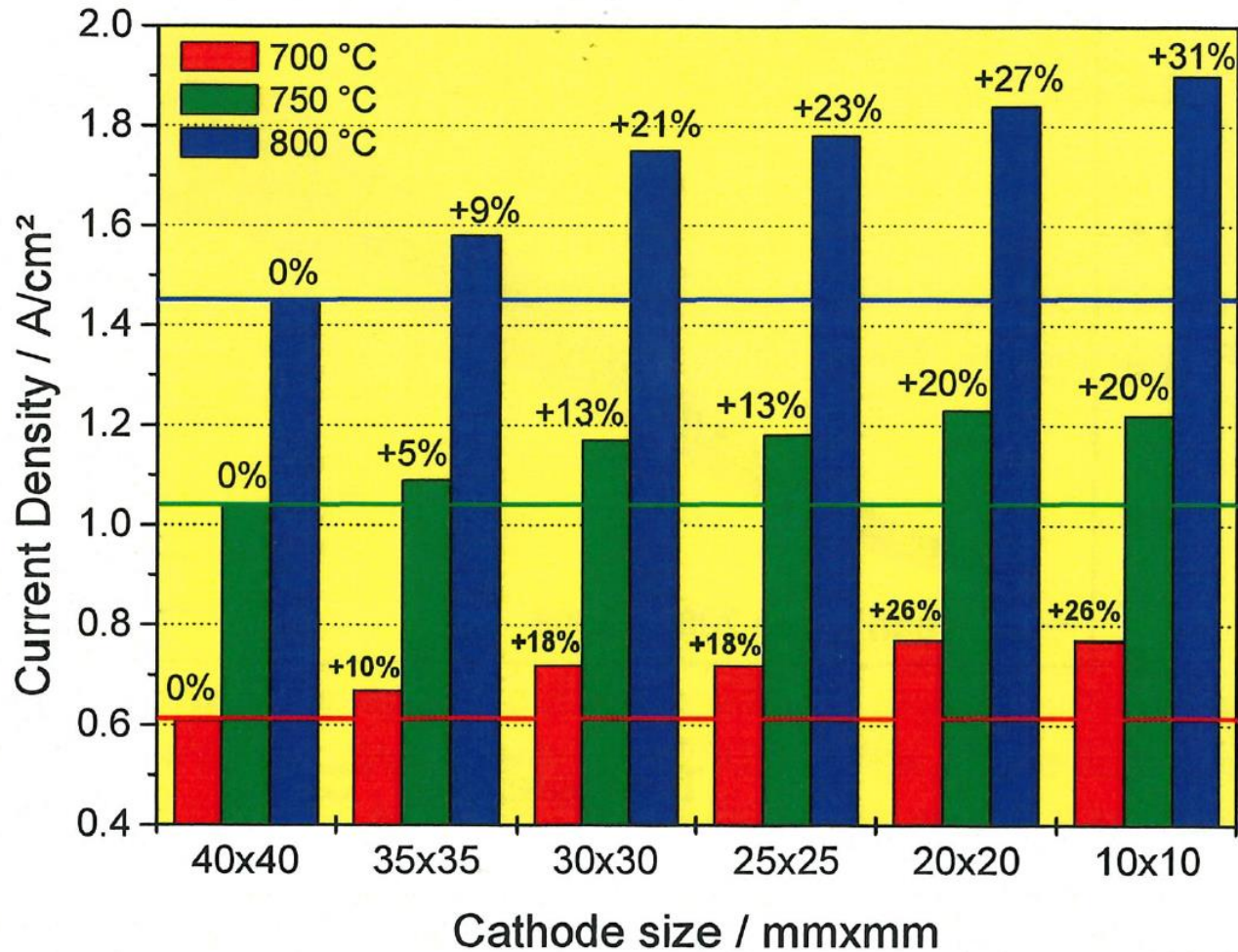
PPD – find the maximum?



- Measuring single cells with 16 cm² electrode area results in currents of 32 A @ 2 A/cm²
- No maximum in power density
- The maximum current depends on the experimental setup, not the cell!
- What if you reduce the active area?

Cell: LSC | 0.5 μm GDC | 2 μm YSZ | Ni-YSZ
A = 16 cm²; 20% H₂O / H₂; air

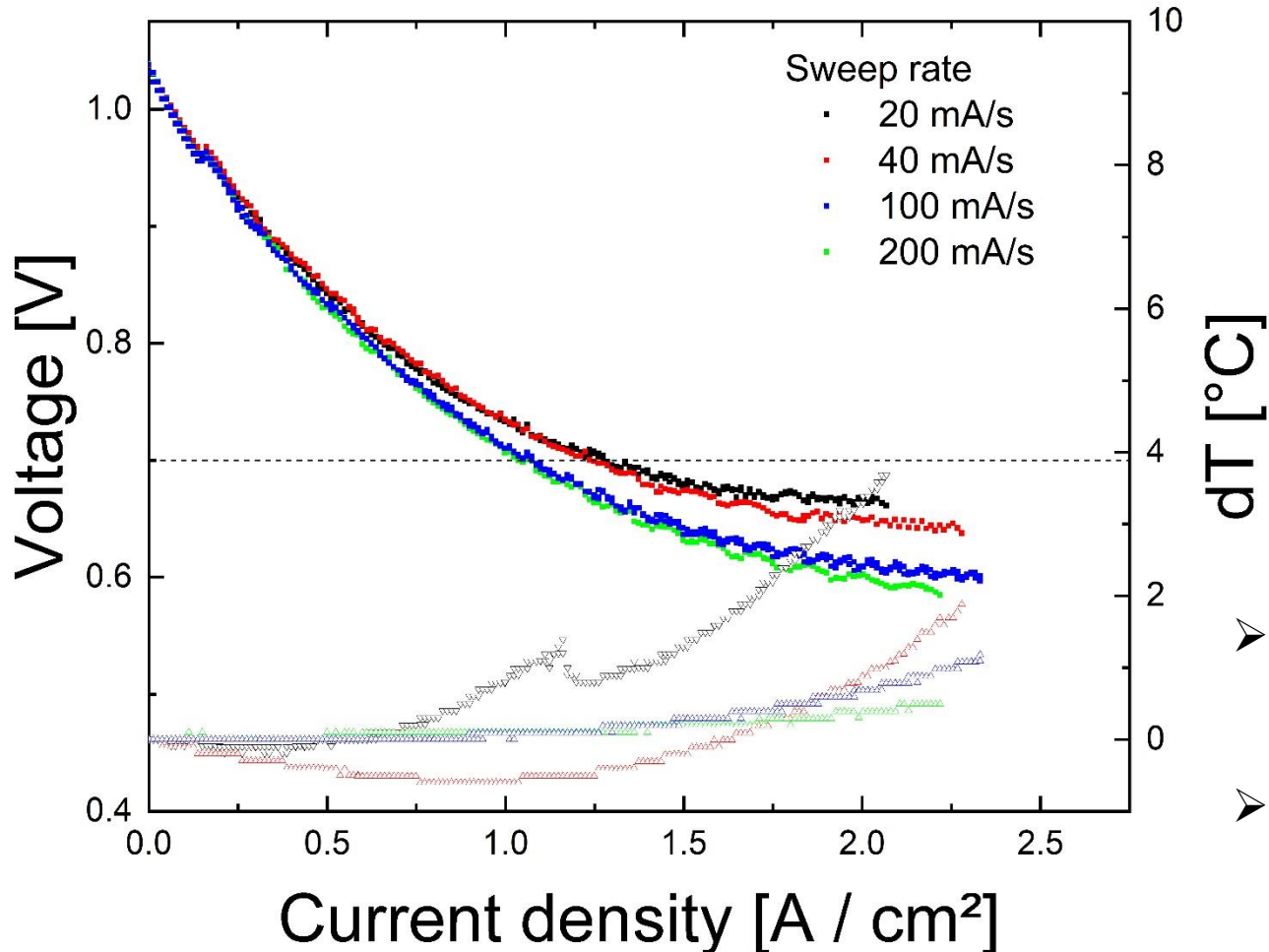
Effect of electrode area



Confidential data from Forschungszentrum Jülich, V. Haanappel

- Electrode-supported cells (Ni-YSZ|YSZ|LSM) contacted with Ni and Pt meshes
- Electrode area varied between 1 cm² and 16 cm²
- Substantial increase in current density (@ 700 mV) with decreasing electrode size
- Difference in power density is likely due to differences in fuel utilization (gas flow rate was constant for all cells)

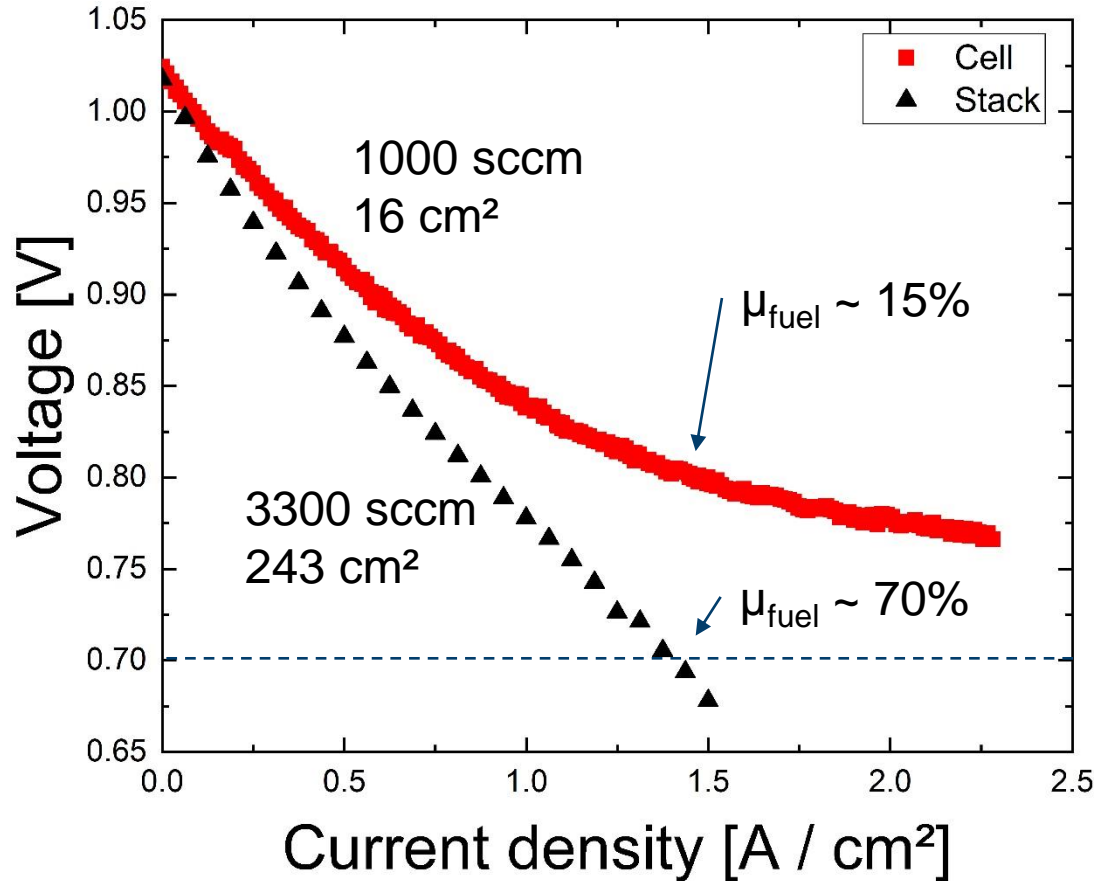
Effect of current sweep rate



Cell: LSC | 0.5 μm GDC | 2 μm YSZ | Ni-YSZ
A = 16 cm²; 20% H₂O / H₂; air; 650 °C

- Current sweep rate has a profound effect on measured performance (~ 20%)
- Probably related to local temperature increase (local temperature \neq thermocouple temperature)

Transfer of cell performance into stacks



Cell: LSC | 0.5 μm GDC | 2 μm YSZ | Ni-YSZ

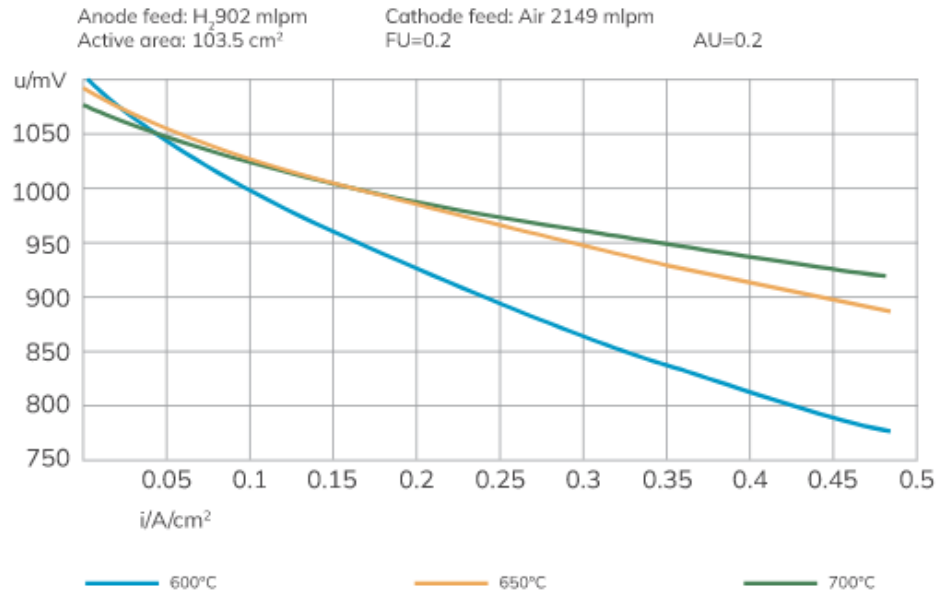
- Integrating cells into stack leads to lower power due to:
 - Contact resistances
 - High fuel utilization
- “Real” operation in a stack makes high current density very difficult to realize (high gas flow rates and pressure drops, large gradients across cell)
- High current density → low system efficiency
- Difference between cell and stack performance increases with higher current density

Cell testing – how does industry do it?

Example: Elcogen (ES)

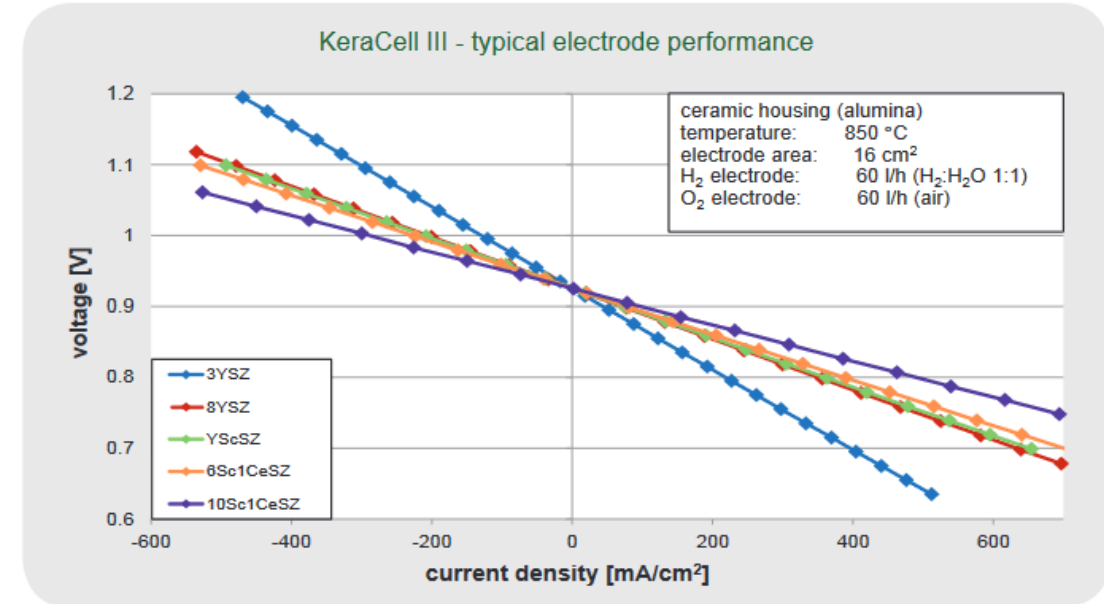
(<https://elcogen.com/products/solid-oxide-fuel-cells/>)

Elcogen Single Cell UI curve comparison



Example: Kerafol (DE)

(https://www.kerafol.com/_wpframe_custom/downloads/files/Kerafol_SOFC_Datenblaetter_KeraCell-III_EN_06-21___143516-14062021.pdf)



- The (few) published metrics from industry reflect the concerns about cell testing. Testing is kept to low current densities.

Problems with PPD - summary

➤ The PPD is typically found at a point that:

- shows low cell efficiency
- is unstable with regards to Ni oxidation
- is not available for every experiments
- depends on cell geometry / test bench characteristics
- Cell performance depends on
 - Gas composition & flow rate
 - Contacting materials
 - Temperature measurement & increase
 - Current sweep rate

Recommended reading:

A. Weber, tm - Technisches Messen, 89 (2022) 97-106.

V.A.C. Haanappel, M.J. Smith, A review of standardising SOFC measurement and quality assurance at FZJ, Journal of Power Sources 171 (2007) 169–178

A better metric

- Variables to define

- Temperature
- Current density / voltage
- Electrode size
- Gas flow rates or gas utilization
- Gas composition

- Variables to eliminate

- Sweep rate -> constant operation
- Open circuit voltage (leakage in test setup) -> use increased humidity

Performance metrics for SOEC

- The situation is better for electrolysis
- The thermoneutral voltage $V_{TN} = \frac{-\Delta H(T)}{nF}$ is the voltage at which the cell neither produces nor consumes heat (under isothermal conditions!)
- Since *Voltage efficiency* $= \frac{V_{TN}}{V_{cell}}$, voltage efficiency is 100% at $V_{cell} = V_{TN}$
- The current density at V_{TN} is therefore a good metric for SOECs
- Other parameters (especially gas composition, also flow rates etc) must be controlled

Summary

- Peak power density (PPD) is extensively used to report cell performance
- However, there are a number of problems with PPD
- Ultimately, it is important to report all significant parameters to enable an interpretation of cell performance
- Implementing better metrics may improve reliability of literature data



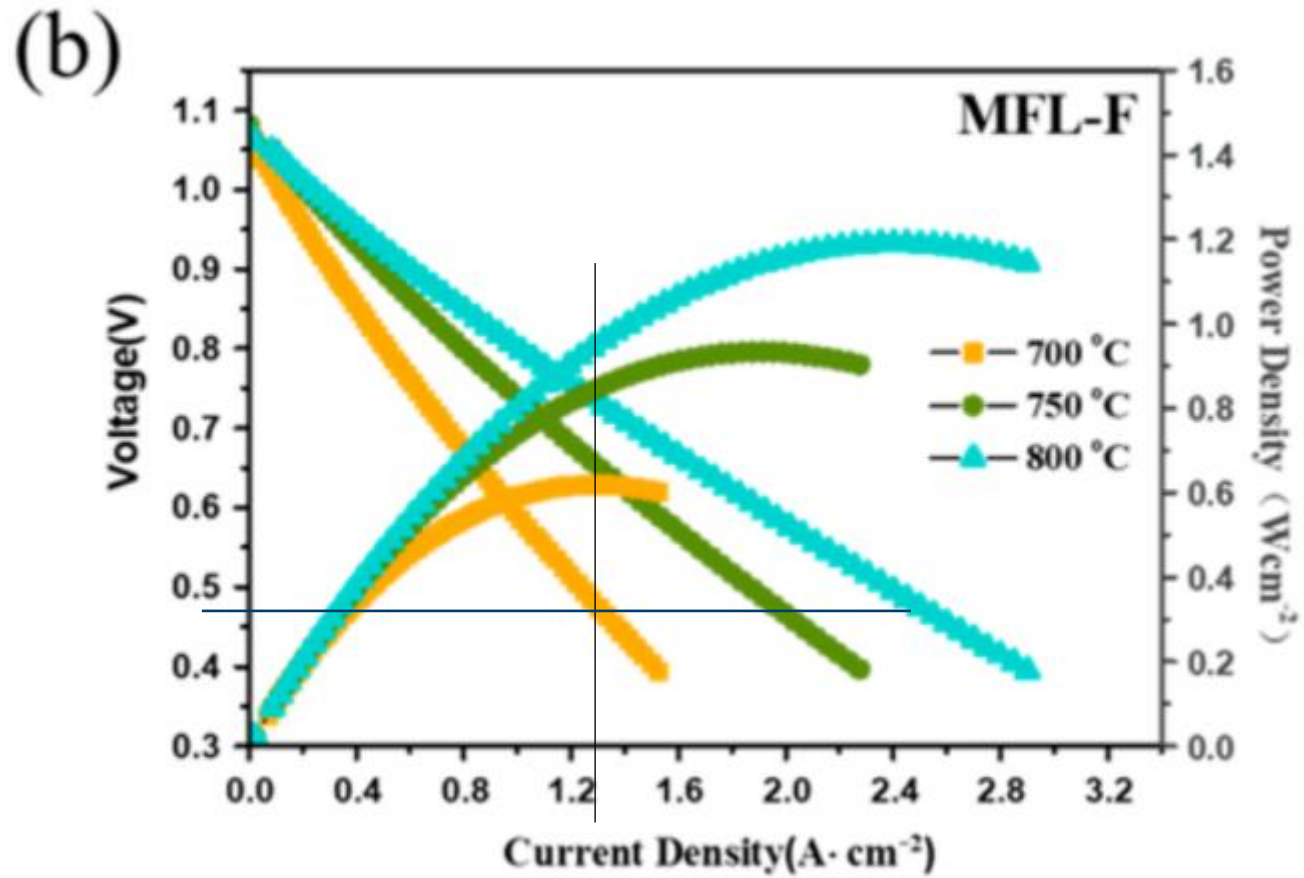
THANK YOU FOR YOUR ATTENTION

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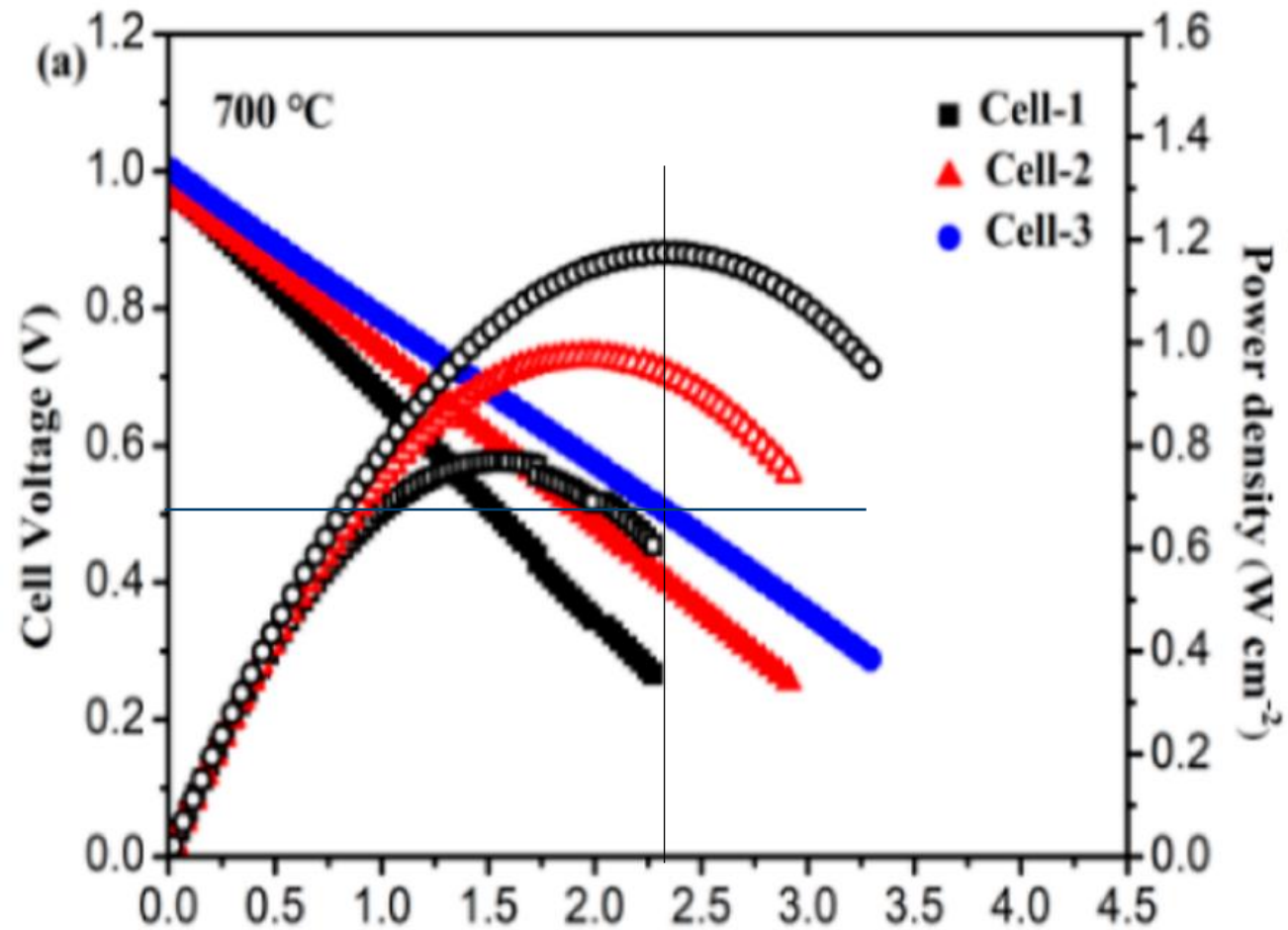
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- [2] Z. Zheng, J. Jing, H. Yu, Z. Yang, C. Jin, F. Chen, S. Peng, ACS Sustainable Chemistry & Engineering, 10 (2022) 6817-6825. **(A)**
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- [1] J. Bai, D. Zhou, X. Zhu, N. Wang, R. Chen, B. Wang, ACS Applied Energy Materials, 5 (2022) 11178-11190. **(B)**
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Ref A



Ref B



Ref C

